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Band of stability. the stable nuclei
cluster over a range of neutron-
proton ratios. Chapter 22 Review:
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SECTION 22-2 OBJECTIVES Define

and relate the terms radioactive decay and nuclear radiation.

Describe the different types of radioactive decay and their effects on the nucleus. Define the term half-life, and explain how it relates to the stability of a nucleus. Define

and relate the terms decay series, parent nuclide, and daughter nuclide. CHAPTER 22 Nuclear

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nuclide, and nuclear reaction.

Summarize the differences between
nuclear reactions and chemical

reactions. Section 22.2 Define
radioactivity and radionuclide. Write

balanced equations for nuclear

reactions, CHAPTER 22. NUCLEAR

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adventure as well as experience

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Answer Key Chapter 22: Nuclear
Chemistry Section 22-1: The

Nucleus • Atomic nuclei= protons
and neutrons (together are
nucleons) o Nuclide= an

atom—identified by # of

protons/neutrons in nucleus Mass

Defect and Nuclear Stability • Mass

defect= difference between mass of
an atom and sum of the masses of
protons/neutrons/electrons o

Caused by conversion of mass to
energy when nucleus forms Nuclear

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Nuclear Chemistry Section 22

... Chapter 22 Review Nuclear

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Answers CHAPTER 22 TEST Nuclear

Chemistry Class MULTIPLE CHOICE

On the line at the left of each

statement, write the letter of the

choice that best completes the

statement or answers the question.

After converting units, the nuclear

mass defect is equivalent to the a.

atomic mass b. electrostatic force c.

energy of chemical reaction

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