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Chapter 11: A new section has been

added on hydrogen bonding and

how the properties of water relate

to this concept. This new section

replaces a deleted section that

dealt with types of solids. • Chapter

12: Within this chapter, the order in

which the various gas laws are

introduced has been altered. All

discussion of STP conditions is now

presented in a section entitled

“molar volume.” Stoker,

Introduction to Chemical Principles |

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concisely represented by a

chemical 1. The substances that

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undergo a chemical change are the
2. The new substances formed in a
chemical reaction are the 3. 05 CTR
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Whether one metal will displace
another metal from a compound
depends upon the relative
reactivities of the 11.2 Types of
Chemical Reactions> Use the
3-step problem-solving approach
you learned in Chapter 1. SECTION
11.1 DESCRIBING CHEMICAL
REACTIONS 1. Write the skeleton
equation for the reaction between
hydrogen and oxygen that produces
water. SECTION 11.1 DESCRIBING
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784. 16.5 PFR/CSTR Series RTD

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Kister and the Two-Parameter

Margules ...

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